

Chapter 12 Stoichiometry Vocabulary Review Answer Key

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Chapter 12 Review Stoichiometry Important Vocabulary • Stoichiometry- The calculations of quantities in chemical reactions in a subject of chemistry • Mole ratio- a conversion factor derived from the coefficients of a balanced chemical equation interpreted in terms of moles

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Chapter 12 REVIEW: Stoichiometry, Theoretical, Actual & Percent yield Part I. Stoichiometry 1. 1 $ZnI_2 \rightarrow 1 Zn + 1 I_2$ How many grams of iodine will you produce if you begin your rxn with 56.7g of zinc iodide? 56.7 g 1 mole 1 mole 253.8 g 319.19 g 1 mole 1 mole = 45.08 grams of Iodine 2. 1 $Pb_2(CO_3)_3 + 6 K \rightarrow 2 Pb + 3 K_2CO_3$

Chapter 12 REVIEW: Stoichiometry, Theoretical, Actual ...

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Chapter 12 Stoichiometry Vocabulary Review Answers

STOICHIOMETRY 12 Vocabulary Review Match the correct vocabulary term to each numbered statement. Write the letter Of the correct term on. the line. Column A 1. the starting materials in a chemical reaction 2. a conversion factor derived from the coefficients of a balanced chemical equation interpreted in terms Of moles

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ACompletion Use this completion exercise to check your understanding of the concepts and terms that are introduced in this section. Each blank can be completed with a term, short ...

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